

Jeroen van Ameijde,^a Andrew R. Cowley,^{b*} George W. J. Fleet,^a Robert J. Nash,^c Michela Iezzi Simone^a and Raquel Soengas^a

^aDepartment of Organic Chemistry, Chemistry Research Laboratory, University of Oxford, Mansfield Road, Oxford OX1 3TA, England, ^bChemical Crystallography Laboratory, Chemistry Research Laboratory, University of Oxford, Mansfield Road, Oxford OX1 3TA, England, and ^cMolecular Nature Ltd, Institute of Grassland and Environmental Research, Aberystwyth SY23 3EB, Wales

Correspondence e-mail:
andrew.cowley@chem.ox.ac.uk

Key indicators

Single-crystal X-ray study
 $T = 150$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.028
 wR factor = 0.034
Data-to-parameter ratio = 7.1

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

5-Amino-5-deoxy-2-C-hydroxymethyl-2,3-O-isopropylidene-D-talono-1,5-lactam

The title compound, $\text{C}_{10}\text{H}_{17}\text{NO}_6$, was prepared by carrying out three $\text{S}_{\text{N}}2$ displacements on a branched sugar derivative, one of which was not planned. Its crystal structure was determined to confirm the identity and stereochemistry of the product.

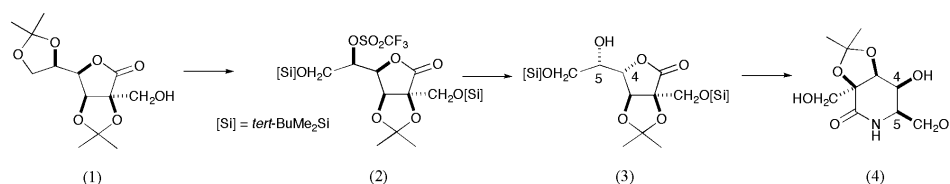
Received 3 September 2004

Accepted 7 October 2004

Online 30 October 2004

Comment

Even though the value of carbohydrate building blocks to provide access to enantiomerically pure synthetic materials is well recognized, there are no easily available branched sugar intermediates (Bols, 1996). The Kiliani reaction on cheap ketoses, although hitherto hardly explored, produces protected branched carbohydrates easily. Such materials are likely to have many uses, but initially we are studying the easy preparation of branched sugar mimetics in which the ring oxygen of the sugar is replaced by nitrogen (Winchester & Fleet, 1992; Asano *et al.*, 2000). The biological properties of branched alkaloids are promising (Ichikawa & Igarashi, 1995; Ichikawa *et al.*, 1998), but the difficulties in the synthesis of such compounds have hindered a substantive investigation of these properties. The branched diacetone (1) was readily prepared from D-fructose and was readily transformed into the trifluoromethanesulfonate (2). It was anticipated that treatment of (2) with an oxygen nucleophile would result in a single inversion of configuration at C5. However, the major product isolated, (3) had undergone inversion of configuration at both C4 and C5. The alcohol (3) was elaborated by standard reactions to the title lactam (4), the structure of which is hereby firmly established by X-ray crystallographic analysis. The configuration at C4 of the lactam (4) unequivocally demonstrates that an unexpected double inversion took place in the transformation of (2) to (3).



The NH and OH groups all form clearly defined intermolecular hydrogen bonds, linking bilayers of molecules running parallel to the crystallographic *ab* plane.

Experimental

The lactam (4) was prepared from the diacetone (1) derived from fructose (Hotchliiss *et al.*, 2004). The title material was crystallized from methanol to yield colourless plates.

Crystal data

C₁₀H₁₇NO₆
M_r = 247.25
 Monoclinic, *P*2₁
a = 6.1266 (2) Å
b = 6.7254 (2) Å
c = 13.8419 (5) Å
 β = 99.6456 (14)°
V = 562.28 (3) Å³
Z = 2

D_x = 1.460 Mg m⁻³
 Mo *K*α radiation
 Cell parameters from 5291 reflections
 θ = 5–28°
 μ = 0.12 mm⁻¹
T = 150 K
 Plate, colourless
 0.38 × 0.38 × 0.14 mm

Data collection

Nonius KappaCCD diffractometer
 ω scans
 Absorption correction: multi-scan
 (DENZO/SCALEPACK;
 Otwinowski & Minor, 1997)
T_{min} = 0.96, *T_{max}* = 0.98
 5291 measured reflections

1372 independent reflections
 1211 reflections with *I* > 3σ(*I*)
R_{int} = 0.030
 θ_{\max} = 27.4°
h = -7 → 7
k = -8 → 8
l = -17 → 17

Refinement

Refinement on *F*²
R = 0.028
wR = 0.034
S = 1.10
 1211 reflections
 170 parameters

H atoms treated by a mixture of independent and constrained refinement
 Weighting scheme: see text
 $(\Delta/\sigma)_{\max}$ = 0.003
 $\Delta\rho_{\max}$ = 0.20 e Å⁻³
 $\Delta\rho_{\min}$ = -0.16 e Å⁻³

Table 1

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1...O6 ⁱ	0.94 (3)	1.96 (3)	2.8510 (19)	156 (3)
O2—H2...O1 ⁱⁱ	0.86 (3)	1.89 (3)	2.7408 (17)	169 (3)
O5—H3...O2 ⁱⁱⁱ	0.91 (4)	1.86 (4)	2.7338 (18)	161 (3)
O6—H4...O5 ^{iv}	0.93 (3)	1.75 (3)	2.6610 (18)	167 (3)

Symmetry codes: (i) $-x, y - \frac{1}{2}, 1 - z$; (ii) $1 + x, y, z$; (iii) $x, 1 + y, z$; (iv) $x - 1, y, z$.

The weighting scheme used a Chebychev polynomial (Watkin, 1994; Prince, 1982): $w = [1 - [(F_o - F_c)/6\sigma(F)]^2]/[1.14T_0(x) + 0.561T_1(x) + 0.916T_2(x)]$, where $x = F_c/F_{\max}$.

Friedel pairs of reflections were merged prior to use in refinement. The absolute configuration of the compound was assumed on the basis of that of the optically pure starting material. The NH and OH H atoms were located in a difference Fourier map and their coordinates and isotropic displacement parameters were subsequently refined. All other H atoms were positioned geometrically (C—H = 1.00 Å), with isotropic displacement parameters set equal to 1.2*U*_{eq} of the parent atom.

Data collection: COLLECT (Nonius, 2000); cell refinement: DENZO (Otwinowski & Minor, 1997); data reduction: COLLECT and DENZO; program(s) used to solve structure: SIR92 (Altomare *et al.*, 1994); program(s) used to refine structure: CRYSTALS (Better-

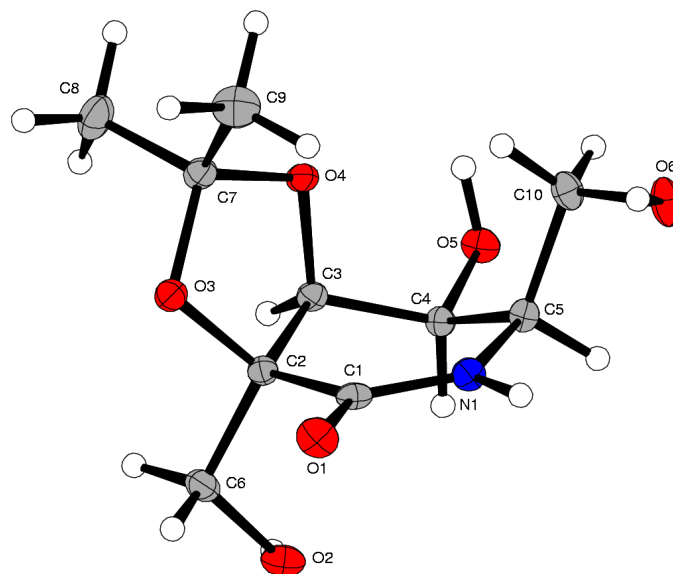


Figure 1

View of the title molecule, showing displacement ellipsoids at the 40% probability level. H atoms are shown as spheres of arbitrary radius.

idge *et al.*, 2003); molecular graphics: ATOMS (Shape Software, 2002); software used to prepare material for publication: CRYSTALS.

Financial support (to RS and MIS) provided through the European Community's Human Potential Programme under contract HPRN-CT-2002-00173 is gratefully acknowledged.

References

- Altomare, A., Casciarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). *J. Appl. Cryst.* **27**, 435.
 Asano, N., Nash, R. J., Molyneux, R. J. & Fleet, G. W. J. (2000). *Tetrahedron Asymmetry*, **11**, 1645–1680.
 Betteridge, P. W., Carruthers, J. R., Cooper, R. I., Prout, K. & Watkin, D. J. (2003). *J. Appl. Cryst.* **36**, 1487.
 Bols, M. (1996). *Carbohydrate Building Blocks*. New York: John Wiley and Sons.
 Hotchkiss, D., Soengas, R., Simone, M. I., van Ameijde, J., Hunter, S., Cowley, A. R. & Fleet, G. W. J. (2004). *Tetrahedron Lett.* **45**. Accepted.
 Ichikawa, Y. & Igarashi, Y. (1995). *Tetrahedron Lett.* **36**, 4586–4587.
 Ichikawa, Y., Igarashi, Y., Ichikawa, M. & Suhara, Y. (1998). *J. Am. Chem. Soc.* **120**, 3007–3018.
 Nonius (2000). COLLECT. Nonius BV, Delft, The Netherlands.
 Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr and R. M. Sweet, pp. 307–326. New York: Academic Press.
 Prince, E. (1982). *Mathematical Techniques in Crystallography and Materials Science*. New York: Springer-Verlag.
 Shape Software (2002). ATOMS for Windows. Version 6.0. Shape Software, 521 Hidden Valley Road, Kingsport, TN 37663, USA.
 Watkin, D. J. (1994). *Acta Cryst.* **A50**, 411–437.
 Winchester, B. & Fleet, G. W. J. (1992). *Glycobiology*, **2**, 199–210.